

RESEARCH ARTICLE

Application of Biosynthesized Silver Nanoparticles Against a Cancer Promoter Cyanobacterium, *Microcystis aeruginosa*

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Abstract

Background: Nanotechnology opens new applications in many fields including medicine. Among all metallic nanoparticles, silver nanoparticles (silver NPS) have proved to be the most effective against a large variety of organisms including toxic cyanobacteria. **Materials and Methods:** Silver NPs were biosynthesized *in vivo* with different alga species namely, *Spirulina platensis*, *Chlorella vulgaris* and *Scenedesmus obliquus* following two scenarios. First: by suspending a thoroughly washed algae biomass in 1 mM aqueous AgNO₃ solution. Second: by culturing them individually in culture media containing the same concentration of AgNO₃. Silver NPs were characterized using UV-Vis spectroscopy, transmission electron microscopy (TEM), energy dispersive analysis (EDX) and Fourier transform infra-red (FTIR) spectroscopy. The biosynthesized silver NPs were tested for cytotoxic activity against a cancer promoter cyanobacterium *Microcystis aeruginosa*, considering effects on cell viability and chlorophyll content. **Results:** The surface plasmon band indicated the biosynthesis of silver NPs at ~400 nm. Transmission electron microscopy (TEM) revealed that the silver NPs had a mean average size below 100 nm. Energy-dispersive analysis X-ray (EDX) spectra confirmed the presence of silver element. FTIR spectral analyses suggested that proteins and or polysaccharides may be responsible for the biosynthesis of silver NPs and (-COO-) of carboxylate ions is responsible for stabilizing them. The toxic potentialities of the biosynthesized silver NPs against the cancer promoter cyanobacterium, *Microcystis aeruginosa* showed high reduction in viable cells count and the total chlorophyll content. **Conclusions:** The potential activity of the biosynthesized silver NPs from the studied algae species against *Microcystis aeruginosa* cells is expected to be mainly mediated by the release of silver ions (Ag⁺) from the particle surface and bioactive compounds as indicated by FTIR analysis.

Keywords: Biosynthesis - *Microcystis aeruginosa* - silver nanoparticles - toxic activities

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Introduction

Cancer is already a major health problem. It is the second main cause of mortalities in the modern world after heart diseases, with more than 10 million new cases every year. This outline is expected to rise in the next few decades. In fact, around one in three people will be diagnosed with cancer throughout their lifetime (Siegel et al., 2012). The lifestyle changes suggest that the burden of neoplasia will become heavier over time, especially with increasing obesity and aging of what are now still youthful populations (Salim et al., 2009). The importance of environmental exposure to contaminants has also been highlighted (Safi, 2002).

Anthropogenic activities along with deficient water management have led to the enhancement of eutrophication in water bodies all over the world (Carmichael, 2007). Eutrophication processes plus specific environmental conditions can lead to cyanobacterial blooms, which are characterized by excessive proliferation of cyanobacterial

cells that produce their toxins (deFigueiredo et al., 2004). Within the large family of cyanobacteria, the cancer promoting, *Microcystis aeruginosa* is the most common bloom-forming species that is able to produce cyanotoxin known as microcystin. Microcystin is one of the best studied classes of cyanobacterial toxins (Ouellette and Wilhelm 2003; Chen et al., 2009).

The risk of exposure to dissolved toxin immediately after the peak of a bloom must be addressed because cyanotoxins can persist even though the bloom has dissipated (Lawton et al., 1994). Prolonged exposure to sublethal doses of microcystin has been epidemiologically linked to primary liver cancer in humans (Yu, 1995).

The liver specificity of Microcystin is due to their selective uptake by hepatocytes through the membrane transport family, Organic Anion Polypeptide Transporters (OAPT) that mediates the uptake and elimination of numerous xenobiotics (Hagenbuch and Meier, 2003; Fischer et al., 2005). Microcystin-LR, inhibits protein phosphatases 1 and 2A (PP1 and PP2A), which are serine/

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threonine phosphatases, sustains the phosphorylation of proteins and induces tumor promotion mediated through signal transduction, resulting in development of tumors. Humans are often exposed to these tumor promoters in the environment, the relationship between human cancer development and the presence of PP1 and PP2A inhibitors is a significant objective of cancer research (Fujiki and Suganuma, 2009).

Besides the liver, the effects of microcystins on other human organs including the kidneys and intestine (Hagenbuch and Meier, 2003), brain (Maidana et al., 2006), lungs (Soares et al., 2007) and reproductive system (Ding et al., 2006) have been investigated.

However, natural products including plant extracts and herbs have been used as medicine sources (Abu-Rabia, 2005; Liu et al., 2012). Nanotechnology opens new applications in many fields including medicine, material science, manufacturing and various technologies. Among all metallic nanoparticles, silver NPs have proved to be the most effective against large variety of organisms including cyanobacteria (Gong et al., 2007). The biological system for biosynthesis of silver NPs must be consisting of environmentally acceptable solvent system, eco-friendly reducing and capping agents (Vigneshwaran et al., 2007; Xie et al., 2007), high-yield, low cost, non-toxic and environmentally benign procedures (Thakkar et al., 2010).

At present, only few studies have investigated the interactions of silver NPs with cyanobacteria. Therefore, the rationale of this work was to investigate algae mediated synthesis of *in vivo* silver NPs, using three selected microalgae species belonging to different groups namely, *Spirulina platensis*, *Chlorella vulgaris* and *Scenedesmus obliquus*. The bio-reduction of AgNO_3 ions in solution was confirmed and fully characterized. The potentialities of biosynthesized silver NPs was determined against the cancer promoter, *Microcystis aeruginosa* which can be further used as a mean of detoxification of water bodies contaminated by this cyanobacterium in swimming pools.

Materials and Methods

Source of algal strains and production of biomass

The cyanobacterium *Spirulina platensis* and two green algae (*Chlorella vulgaris* and *Scenedesmus obliquus*) were kindly provided from Phycology Laboratory, Faculty of Science, Alexandria University. *Microcystis aeruginosa* was obtained from Phycology laboratory, Faculty of Science, Tanta University, Egypt. *S. platensis* was cultured in BG-11 medium for 15 days. Green algae were cultured in a modified Bold Basal medium (Jena et al., 2013). For the production of biomass, exponentially growing algae culture was transferred into fresh sterile medium [10% (v/v) of inoculums], incubated at $28 \pm 2^\circ\text{C}$ and day/night rhythm (16/8), respectively. *M. aeruginosa* was cultured in 1 liter conical flasks containing 400 mL medium (Allen's and Stanier, 1968), and kept in controlled conditions of continuous light (45 $\mu\text{mol/ms}$) at $25 \pm 2^\circ\text{C}$.

In vivo Synthesis of silver NPs and their characterization

There were two methods used for the *in vivo* synthesis of silver NPs by algae cultures. The first method: algal

cells from logarithmic phase were centrifuged at 4000 rpm for 20 min. The supernatant was removed and the biomass was washed with sterile deionized water to remove traces of media salts leftovers. The washing was repeated for five times (Jenal et al., 2013). The silver NPs were prepared by taking 5 gm of algae biomass from an exponential growth phase in a 250 ml Erlenmeyer flask with 95 ml of 1 mM aqueous AgNO_3 solution (pH 7) (Sudha et al., 2013). According to the modified method of Devina Merin et al. (2010), the different algae species were individually cultured along with 1 mM AgNO_3 solution and kept in the previously mentioned conditions for two weeks.

Characterization of silver NPs

The properties and structure of silver NPs have been characterized by means of vis-UV spectra of the solution, Transmission Electron Microscope (TEM) and Energy-Dispersive analysis X-ray (EDX) spectrum as well as Fourier Transform Infrared Red spectroscopy (FTIR) spectral Analysis.

The bio-reduction of pure Ag^+ ions in aqueous solution was detected by sampling of aliquots (0.2 mL) of the colloidal suspension, and then diluting the samples with 2 mL deionized water and subsequently measuring Vis-UV spectra of the resulting diluents using UV-6800 UV-Vis Spectrophotometer (JENWAY-Germany). The absorption maxima were scanned at the wavelength of 200-800 nm.

TEM analysis was employed to visualize the size and shape of silver NPs. TEM micrographs were taken by analyzing the prepared grids on (JEOL 100CX, Japan) at the Electron Microscope Unit- Faculty of Science- Alexandria University.

The presence of elemental silver as well as the other elementary compositions of the silver NPs was detected. Energy-dispersive analysis X-ray (EDX) analysis was carried out at 20 KV by X-ray micro-analyzer (Module Oxford 6587INCA X-sight) attached to JEOL 100CX Scanning electron microscopy.

Samples of the aqueous solution of the silver NPs were prepared by centrifugation at 10,000 rpm for 30 min. The pellet was lyophilized and subjected to FTIR analysis by KBR pellets (FTIR grade) method (Kasthuri et al., 2009). The spectrum was recorded in the range of 500 to 4000 cm^{-1} .

Cytotoxic potentials of the biosynthesized silver NPs against *M. aeruginosa*

Aliquots of 30 mL of the *M. aeruginosa* culture in growing media, Allen's and Stanier (1968) were exposed to 0.1, 1 and 10 mgL^{-1} of the biosynthesized silver NPs and kept in controlled conditions of continuous light (45 $\mu\text{mol/ms}$) and temperature ($25 \pm 2^\circ\text{C}$) for 24h. Cell viability and total chlorophyll were determined. The experiments were conducted in triplicate and results are shown as the mean with standard deviations.

Optical microphotographs of *M. aeruginosa*

Morphological changes in *M. aeruginosa* culture exposed to 0.1, 1.0 and 10 mg/L of silver NPs for 24h were determined using a Reichert microscope (Austria- Nr. 365475) and recorded with a Pixelink camera.

Determination of total chlorophyll

Total chlorophyll was measured according to Tandeau de Marsac (1977); 2 ml of the culture were centrifuged at 15,000 g for 10 min in a 2 ml reaction tube. Then 1900 μ l of the supernatant were removed and the cells were resuspended using an ultrasonic bath. Afterwards 900 μ l of 100% methanol were added and incubated in the dark at 4°C for 1 hour. Samples were then centrifuged at 15,000g for 10 minutes and absorption was measured at 650 nm. Chlorophyll concentration in μ g/ml was calculated by the following formula: Chlorophyll (μ g/ml) = (Abs 650 \times 13.9) / 2ml.

Evaluation of algal culture viability

The *M. aeruginosa* cells viability was determined using Olympus BX41 fluorescence Microscope-America. Pictures were recorded using the USB camera with image capture software that is available for recording images.

Results

Biosynthesis of silver NPs

The biosynthesized silver NPs using the different microalgae strains were confirmed by the UV-Vis spectral analysis at various nm. The colour change was due to excitation of Surface Plasmon Vibration which indicated the formation of silver NPs (Figure 1).

Characterization of silver NPs

UV-Visible (UV-Vis) spectral analysis: The Surface Plasmon band indicated the production of silver NPs at \sim 400 nm for *S. platensis* and *C. vulgaris* as well as *Sc. obliquus*, respectively. These results revealed that the silver NPs were dispersed in the aqueous solution with no evidence for aggregation (Figure 1).

Transmission electron microscope (TEM): The biosynthesis of silver NPs was further confirmed by the structural view and determination of the size of nanoparticles under the TEM as shown (Figure 2). Most of the biosynthesized silver NPs seem to be spherical in

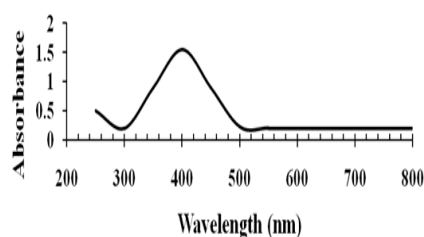


Figure 1. UV-Vis Spectrum of Plasmon Resonance of the Biosynthesized Silver NPs

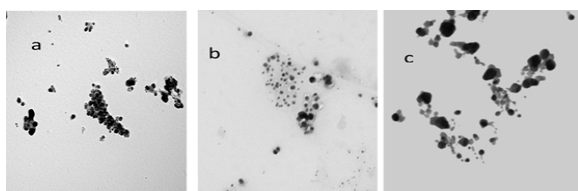


Figure 2. Transmission Electron Micrographs of the biosynthesized silver NPs using *S. platensis*, a) *C. vulgaris*; b) and *Sc. obliquus*; c) respectively (magnification: 10000x)

morphology and well distributed with the mean average size of 20.8 ± 4 , 8.2 ± 3 and 8.8 ± 2 nm for *S. platensis*, *C. vulgaris* and *Sc. obliquus*, respectively.

Energy-dispersive analysis X-ray (EDX) spectrum:

The presence of elemental silver is further confirmed by Energy-dispersive analysis X-ray (EDX) spectra with the absorption peak in the range of 3 to 4 keV as outlined in (Figure 3). However, some of the additional peaks for C, Cu, N, O, P, Mg, S, and Ca were observed.

Fourier Transform Infra-red (FTIR) analysis:

FTIR analyses were carried out to identify the possible biomolecules responsible for the reduction of silver ions and the capping of the bio-reduced silver NPs synthesized by different micro algal species. Figure 4(a) shows FTIR spectrum of silver NPs synthesized using *S. platensis*. The peak at 3396 cm^{-1} is mainly due to N-H stretching vibration presence of secondary amines (protein, lipid), the peak at 1639 cm^{-1} can be assigned to the protein amide I band, mainly $\nu(\text{C}=\text{O})$ stretching and may be due to the N-H bending vibration present in the carbonyl β unsaturated

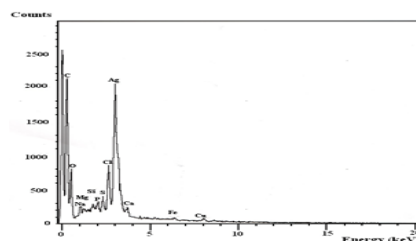


Figure 3. EDX Monographs of the Biosynthesized Silver NPs Showing Elemental Silver in High Signals

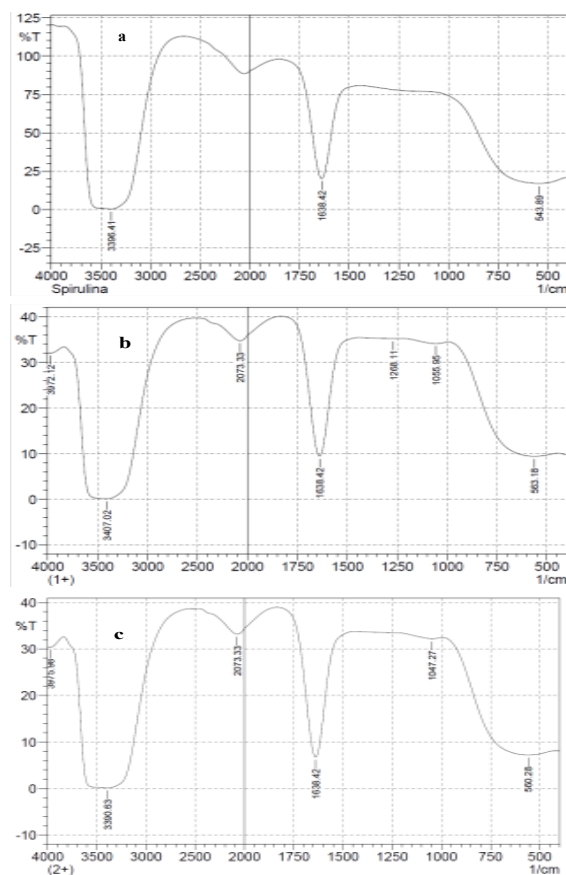


Figure 4. FTIR Spectrum of Protein-Capped Silver NPs Synthesized using *S. platensis*. a) *C. vulgaris*; b) and *Sc. obliquus*; c) respectively

ketone amide and the peak at 544 cm⁻¹ may be due to C-I stretching vibration presence of Iodo compounds.

Figures 4(b) and 4(c) show FTIR spectrum of silver NPs synthesized using *C. vulgaris* and *Sc. obliquus*. Stretching vibrations of proteins were observed by two clear bands. The peaks at 3390 and 3407 cm⁻¹ correspond to Protein ν (N-H) stretching (amide A), for *C. vulgaris* and *Sc. obliquus*, respectively. The other peak at 1638 cm⁻¹ presented in the 2 spectra is due to Protein amide I band mainly ν(C=O) stretching. However the peak at 2073

cm⁻¹ may be assigned to Transition metal carbonyls. The band at 1047 cm⁻¹ is due to Symmetric C-H Stretching vibration, presence of Antioxidant enzyme, Carbohydrate ν(C-O-C) of polysaccharides. However, the peak observed at 1268 cm⁻¹ may be assigned to the presence of C-O asymmetric C-O-C stretching presence of esters. The band at 1055 cm⁻¹ corresponds to Carbohydrate ν(C-O-C) of polysaccharides. Peaks at 560 cm⁻¹ and 563 cm⁻¹ represents C-I stretching vibration presence of Iodo compounds.

The cytotoxic activity of the biosynthesized silver NPs M. aeruginosa cells

Cell viability assay: Cultures of the tumor promoting cyanobacterium, *M. aeruginosa* exposed to different concentrations of the biosynthesized silver NPs for 24h showed cell aggregates' formation compared to control (Figure 5). Large aggregates were observed in the algal culture under the stress of 10 mgL⁻¹ for all the biosynthesized silver NPs.

The percent reduction of cyanobacteria cells' viability, reveal a reduction in viable cells after 24 h treatment. Growth of *M. aeruginosa* under the stress of 1 to 10mgL⁻¹ biosynthesized silver NPs resulted in a significant reduction of viable cells (Figures 6A & 6B). The exposure to 1mgL⁻¹ silver NPs' induced a % 30.3±0.4, %44.7±0.5 and a % 51.1±0.4 decrease of viable cells, for those silver NPs produced by *S. platensis*, *C. vulgaris*, and *Sc. obliquus*, respectively. As the concentration of silver NPs increased, the cyanobacteria cells' viability was decreased. High reduction of the viable cell count was achieved by using 10 mgL⁻¹ of the biosynthesized silver NPs. The percentages of reduction of viable cells reached % 92±2.0, % 96±3.0 and % 98±2.0 for the silver NPs produced by *S. platensis*, *C. vulgaris* and *Sc. obliquus*, respectively.

Effect of silver NPs on chlorophyll content

The reduction levels in chlorophyll content of the tumor promoting cyanobacterium cells exposed to different concentrations of the biosynthesized silver NPs for 24 hours are presented in Figure 7. The results show that using silver NPs produced by *S. platensis*, *C. vulgaris* and *Sc. obliquus*, have reduced the chlorophyll content of cyanobacterium by % 20.3±0.4, % 34.7±0.3, and % 41.1±0.5 respectively. Increasing the silver NPs

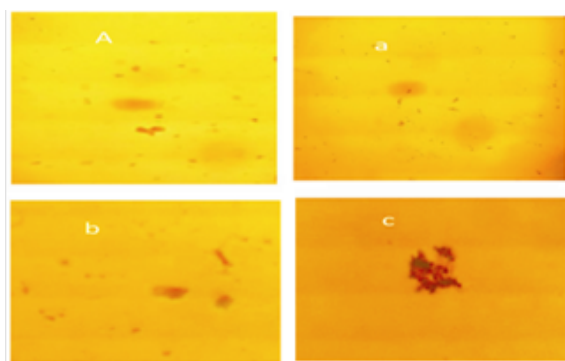


Figure 5. Morphological Changes of *M. aeruginosa* Colonies Showing Control Cells (A) and Images of Cells Exposed to 10 mgL⁻¹ of Silver NPs Synthesized using *S. platensis*, a; and *C. vulgaris*, b; as well as *Sc. Obliquus*, c; Respectively

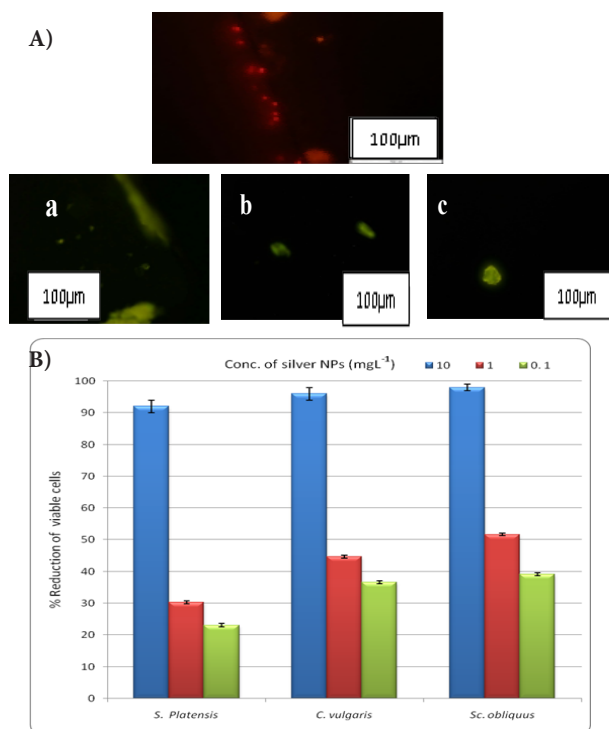


Figure 6. A) Images of Cells Showing Control *M. aeruginosa* Cells (A) and Images of Cells Exposed to 10 mgL⁻¹ of Silver NPs Synthesized using *S. platensis*, a; *C. vulgaris*, b; and *Sc. obliquus*, c; respectively. [Red fluorescence is caused by chlorophyll and represents viable cells. Non-viable cells show green unspecific auto fluorescence]. B) Percent Reduction of Viable Cells of *M. aeruginosa* Under the Stress of Different Concentrations of Silver NPs for 24h

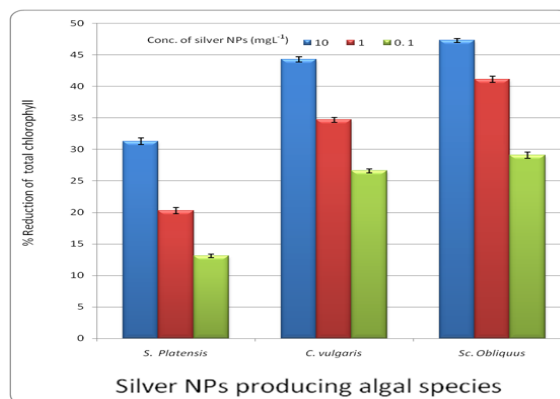


Figure 7. Percent Reduction of Chlorophyll Content of *M. aeruginosa* Under the Stress of Different Concentrations of Silver NPs for 24h

concentrations have caused more pronounced decrease in the chlorophyll content. The application of silver NPs (10.0 mgL^{-1}) produced by *Sc. obliquus* led to the highest reduction of the total chlorophyll content of *M. aeruginosa* by $47.3\% \pm 0.3$.

Discussion

Due to their unique physical and chemical properties, nanomaterials are extensively studied. Natural nanoparticles (NP) were investigated for their physicochemical behavior in the environment (Buffle, 2006; Novak and Bucheli, 2007), but less for human health and environmental consequences with the exception of small-sized dust particles (Yin et al., 2013).

As observed in this study one of the possible roles of the microorganisms is providing a multitude of nucleation centers; establishing conditions for obtaining highly dispersed nanoparticle systems. In addition, using microorganisms in green biosynthesis of nanoparticles slow down aggregation, or entirely prevent it by immobilizing the particles and providing viscous medium (Sun and Xia, 2002). Thus the biosynthesized nanoparticles have highly intricate architectures and are ordered during assembly.

The biosynthesized silver NPs were monitored by UV-Vis spectrophotometer at various nm-s. The surface Plasmon band in the silver NPs solution remained close to 400 nm throughout the reaction period indicating that the particles are dispersed in the aqueous solution, with no evidence for aggregation. The silver ions are reduced by the extracellular reductase enzymes produced by the microorganisms to silver metal in nanometer range (Sosa et al., 2003; Fu et al., 2006). The exact mechanism leading to the formation of silver NPs by the algal biomass is not fully understood; there are still several possible mechanisms involved in the process. It is thought that the first step involves the trapping of metal ions on the surface of algal cells, possibly via electrostatic interaction between the ions and negatively charged carboxylate groups present in the cell surface. Thereafter, the ions are reduced by the enzymes, leading to the formation of nuclei, which subsequently grow through the further reduction of metal ions and accumulation of these nuclei (Mandal et al., 2006).

TEM images of the different silver NPs reveal that the silver NPs seem to be spherical in morphology and well distributed with an average size of 20.8 ± 4.0 ; 8.2 ± 3.0 and 8.8 ± 2.0 nm for *S. platensis* and *C. vulgaris* as well as *Sc. obliquus*, respectively. Similar results were reported by Jena et al. (2013).

The EDX spectrum of the biosynthesized silver NPs confirmed the presence of elemental silver. The presence of an optical absorption band at 3 KeV reveals the presence of pure metallic silver NPs. The other peaks may be due to proteins and cell biomass trapping the silver NPs (Jain et al., 2011).

FTIR measurements were carried out to identify the biomolecules which may be responsible for synthesis and stabilization of silver NPs. FTIR analyses revealed the presence of molecules like proteins, lipids and

carbohydrates. Stretching vibrations of proteins were observed by strong bands in the spectrum of silver NPs synthesized by the three algae species. In accordance with these results, Sable et al. (2012) stated that representative spectra of biosynthesized silver NPs manifest absorption peaks of respective functional groups and indicated the presence of stabilized protein molecules. Moreover, Govindaraju et al. (2008) suggested the interaction of single cell protein of *S. platensis* with aqueous silver nitrate (AgNO_3) for the synthesis of silver NPs. The conformation of protein molecules plays an important role in silver NPs synthesis and stabilization. The results suggested that the capping ligand of the silver NPs may be an aromatic compound. It was reported that the extract of unicellular green algae *C. vulgaris* was used to synthesize single-crystalline silver nanoplates at room temperature (Xie et al., 2007). Proteins in the extract provide dual function of Ag^+ reduction and shape-control in the synthesis of the silver NPs. Up to the current knowledge, this is the first report that concluded the biosynthesis of silver NPs by *Sc. obliquus*.

It was concluded from protein assay of microorganisms that the preparation of silver NPs is a NADH-dependent reductase. The reductase enzyme gains electrons from NADH and oxidizes it to NAD^+ . The enzyme is then oxidized by the simultaneous reduction of silver ions forming silver metal in nano form. In some cases a nitrate-dependent reductase is responsible for the bioreduction process, therefore a complex electron shuttle materials may be involved in the biosynthesis process (Moghaddam, 2010). Moreover, (-COO-) of carboxylate ions is responsible for stabilizing the silver NPs. It was reported that extracts from microbes act both as reducing and capping agents in metal synthesis of nanoparticles. Reduction of metal ions by combinations of biomolecules such as enzymes or proteins, amino acids, polysaccharides, and vitamins (Collera-Zuniga et al., 2005) which may be used as reductants to react with silver ions, leading to silver NPs synthesis in solutions (Li et al., 2007) is environmentally benign, yet chemically complex. The overall suggestions are that proteins and polysaccharides were involved in the reduction of AgNO_3 to form silver NPs and the capping agents may be the aromatic compounds.

The cytotoxicity of silver NPs is expected to be mediated mainly by the release of silver ions (Ag^+) from the particle surface. Consequently, silver NPs can be considered as a source of toxic Ag^+ , which are adsorbed to particle surfaces or formed upon oxidative dissolution in presence of oxygen, ligands, or organisms (Navarro et al., 2008; Xiu et al., 2011).

In the present study, the cultures of the cancer promoter cyanobacterium, *M. aeruginosa* exposed to different concentrations of the biosynthesized silver NPs for 24h showed formation of cell aggregates comparing to control cultures. Large aggregates were observed in the algal cultures under the stress of 10 mgL^{-1} after administration of all biosynthesized silver NPs. In this respect, it was reported that metal nanoparticles like TiO_2 NPs were able to interact directly with the cells' surface through adsorption to the cell walls (Sadiq et al., 2011). The pores

across the cell wall have diameters ranging from 5 to 20 nm that determines its sieving properties, the biosynthesized silver NPs have diameters within the mentioned range. It was suggested that the formation of aggregates might inhibit cells' growth (Navarro et al., 2008). The strong reduction of viable cells count and chlorophyll content further support this suggestion. Cyanobacterium cells' viability, evaluated by fluorescent microscope, revealed a reduction in viable cell number after 24 h treatment. As the concentration of the biosynthesized silver NPs was increased, the algal cells' viability was sharply decreased. The percent reduction of viable cells reached 98 % for the silver NPs produced by *Sc. obliquus*.

The chlorophyll concentration is often used as a measure for the viability of cyanobacteria cultures (Schulze et al., 2011). The data of the present study have shown an expected dose dependent decrease of the chlorophyll concentrations for all the samples along with decreased numbers of viable cells due to silver NPs administration. It was established that chlorophyll auto fluorescence could be accepted as the most convenient method for differentiation between living and dead cells. The tests based on the fluorescence of the dead cell cytoplasm appeared to be useful for vitality assessment, because chlorophyll fluorescence was lost and did not mask cytoplasmic fluorescence (Pouneva 1997). Interestingly, increasing the concentrations of silver NPs have caused more decreasing in the chlorophyll content of the studied Cyanobacterium. The chlorophyll concentration also matched the results of the viability analysis via fluorescence. These findings are in agreements with those of Schulze et al., (2011).

In line with the present results, many studies have investigated the cytotoxic effects of metal nanoparticles against different human cancer cell lines (Rosarin et al. 2013). El-Kassas and Attia, (2014) reported the cytotoxic activity of biosynthesized silver nanoparticles with an extract of the red seaweed *Pterocladia capillacea* on the HepG2 cell line. In addition El-Kassas and El-Sheekh, (2014), studied the cytotoxic activity of gold nanoparticles of *Corallina officinalis* on human breast cancer (MCF-7) cell line. Other investigations are necessary to improve the particle size and other necessary features of the biosynthesized nanoparticles. The results of this work provided strong evidence for the consideration of silver nanoparticles (AgNPs) as antialgal agent against the liver inducing cancer *M. aeruginosa*. Such positive environmental and toxicological applications will be imperative to ensure the nanomaterials design process yields both effective and safe technologies for prevention of liver cancer in human as results of the toxin excreted by *M. aeruginosa*.

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